

Communications

An Efficient Synthesis of the Enantiomeric 17-Phosphasteroid System¹

Summary: Enantiomeric *trans*-benzylphenyl[β -(carbomethoxy)vinyl]phosphine oxide (6) undergoes regio- and stereoselective cycloaddition to 1-vinylnaphthalene (5), providing two diastereomeric adducts 7a and 7b, which are further transformed into optically pure 17-phosphasteroid systems by intramolecular Dieckmann-type condensation.

Sir: A wide variety of heterosteroids isoelectronic with the natural steroidal hormones have been synthesized in the last decade as part of an intensive search for new compounds with potential specific pharmacological properties.²

We have recently contributed to this field with a new synthetic method for the 3-oxophospholane oxide unit which has proven useful in the preparation of the 17-phosphasteroid system³ (Scheme I).

In order to provide an efficient route to enantiomeric 17-phosphasteroids, we have devised a modified version of this approach. The vicinal methoxycarbonyl and benzylphosphoryl functions, which are required for formation of the five-membered ring, are simultaneously introduced to the carbocyclic skeleton by a [2 + 4] cycloaddition process (Scheme II) instead of the nucleophilic 1,4-addition originally employed (Scheme I). Scheme II presents a straightforward example of this new route.

Optically pure *trans*-benzylphenyl[β -(carbomethoxy)vinyl]phosphine oxide of S_P configuration (6),⁴ mp 167.5 °C, $[\alpha]_D^{20} +20.1^\circ$ (*c* 8.1, CHCl₃), was reacted with 1-vinylnaphthalene (5) in a sealed tube at 120 °C (40 h, toluene, CF₃COOH) and afforded a mixture of two diastereomeric cycloadducts 7a and 7b in a ratio 65:35 in 70% yield.^{5,6} Fractional crystallization of this mixture from CCl₄/C₆H₆/CHCl₃ (10:10:1) yielded the minor component 7b: mp 226 °C; $[\alpha]_D^{20} +2.5^\circ$ (*c* 4.2, CHCl₃); ³¹P NMR (CHCl₃) 39.78 ppm. The major component 7a precipitated from the mother liquor after addition of hexane. 7a: mp 183–185 °C; $[\alpha]_D^{20} -19^\circ$ (*c* 5.3, CHCl₃); ³¹P NMR (CHCl₃) 40.36 ppm.

¹³C NMR spectra of both components revealed coupling between ring C-11 carbons (steroid numbering) and

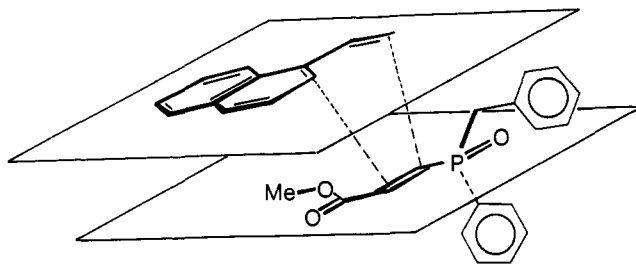


Figure 1.

phosphorus to be 10 and 9.7 Hz for 7a and 7b, respectively. These values are clearly compatible with a three-bond C-P coupling^{8,9} and imply a uniform regioisomerism for 7a and 7b pictured in Scheme II. Similarly, small values of ³J_{P-C(O)} observed for 7a and 7b, i.e., 4.4 and 2.0 Hz, respectively, are completely in accord with the expected *trans* dipseudo-equatorial disposition of the phosphorus and carbomethoxy groupings in both isomers.^{3b,10}

The assignment of the relative and absolute configurations to the isomers 7a and 7b follows from a single-crystal X-ray diffraction analysis of racemic counterpart of 7a¹¹ and from the fact that the precursor 6 was of the known absolute configuration (S_P). These data¹¹ indicate unambiguously the S_P, R_{C-13}, S_{C-14} configuration for the isomer 7a. Accordingly, the minor isomer 7b has to possess the S_P, S_{C-13}, R_{C-14} absolute configuration.

The stereo- and regiochemistry of the performed [2 + 4] cycloaddition may be accounted for simply. The *trans* configurations of 7a and 7b reflect the *trans* configuration of the dienophile 6, whereas the observed regiospecificity is most likely a result of the strong directing effect of the carbonyl group¹² which apparently dominates that of the phosphoryl.¹³ This observation is also suggested by the results of Kashman and Awerbouch.¹⁴

Formation of the two diastereomeric adducts 7a and 7b in unequal amounts argues in favor of asymmetric induction. If in the reactive conformation of the dienophile 6, the conjugated C=O, C=C, and P=O groupings are assumed to be *trans* coplanar, the configurations of two newly created centres of asymmetry, i.e., C-13 and C-14 should be controlled mainly by steric interactions between the phosphorus substituents and the diene¹⁵ coming in endo with respect to the carbomethoxy grouping.¹⁶ In

(1) Paper presented at the International Symposium on Phosphorus Chemistry Directed Towards Biology, Sept. 25–28, 1979, Burzenin, Poland.

(2) (a) Huisman, H. O. *MTP Int. Rev. Sci., Org. Chem.* 1973, 8, 235. Huisman, H. O.; Speckamp, W. N. *MTP Int. Rev. Sci., Org. Chem., Ser. Two* 1976, 8, 207. Huisman, H. O. *Bull. Soc. Chim. Fr.* 1968, 13. (b) Pitt, C. G.; Friedman, A. E.; Rector, D.; Wani, M. C. *Tetrahedron* 1975, 31, 2369. Barcza, S.; Hoffman, C. W. *Ibid.* 1975, 31, 2363. (c) Radhakrishna, A. S.; Berlin, K. D.; Helm, D. v. d. *Pol. J. Chem.* 1980, 54, 495. (d) Symmes, C., Jr.; Quin, L. D. *J. Org. Chem.* 1979, 44, 1048.

(3) (a) Bodalski, R.; Pietrusiewicz, K. M. *Tetrahedron Lett.* 1972, 4209. (b) Pietrusiewicz, K. M. Ph.D. Thesis, Polish Academy of Sciences, Łódź, Poland, 1976.

(4) The synthesis of enantiomeric benzylphenyl[β -(carbomethoxy)vinyl]phosphine oxide from methyl β -chloroacrylate and ethyl benzylphosphinite in three steps and the assignment of its absolute configuration through chemical correlation will be reported elsewhere; Bodalski, R.; Koszuk, J., manuscript in preparation.

(5) All compounds gave correct C, H, P analyses and their ¹H, ¹³C, and ³¹P NMR spectra are in accord with the proposed structures.

(6) The ratio of 7a and 7b was determined from integration of the corresponding ³¹P NMR resonances.

(7) The new sign convention is used.

(8) Quin, L. D. "The Heterocyclic Chemistry of Phosphorus"; Wiley-Interscience: New York, 1981; Chapter 6.

(9) The observed magnitudes of P-C couplings do not fall in the range of 0–3 Hz which is common for ⁴J_{PC}. See: (a) Quin, L. D.; Gordon, M. D.; Lee, S. D. *Org. Magn. Reson.* 1974, 6, 503. (b) Quin, L. D.; Mesch, K. A.; Bodalski, R.; Pietrusiewicz, K. M. manuscript submitted for publication.

(10) Kingsbury, C. A.; Thoennes, D. *Tetrahedron Lett.* 1976, 3037.

(11) Główska, M. L.; Galdecki, Z. *Acta Crystallogr., Sect B* 1981, 273.

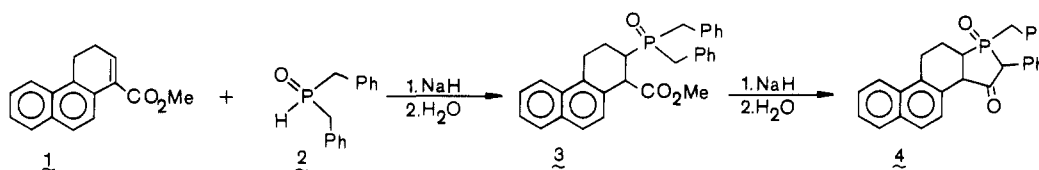
(12) Fleming, I.; "Frontier Orbitals and Organic Chemical Reactions"; Wiley: London, 1976; Chapter 4. Eisenstein, O.; Lefour, J. M.; Anh, N. T.; Hudson, R. F. *Tetrahedron* 1977, 33, 523.

(13) Daniewski, W. M.; Griffin, C. E. *J. Org. Chem.* 1966, 31, 3236.

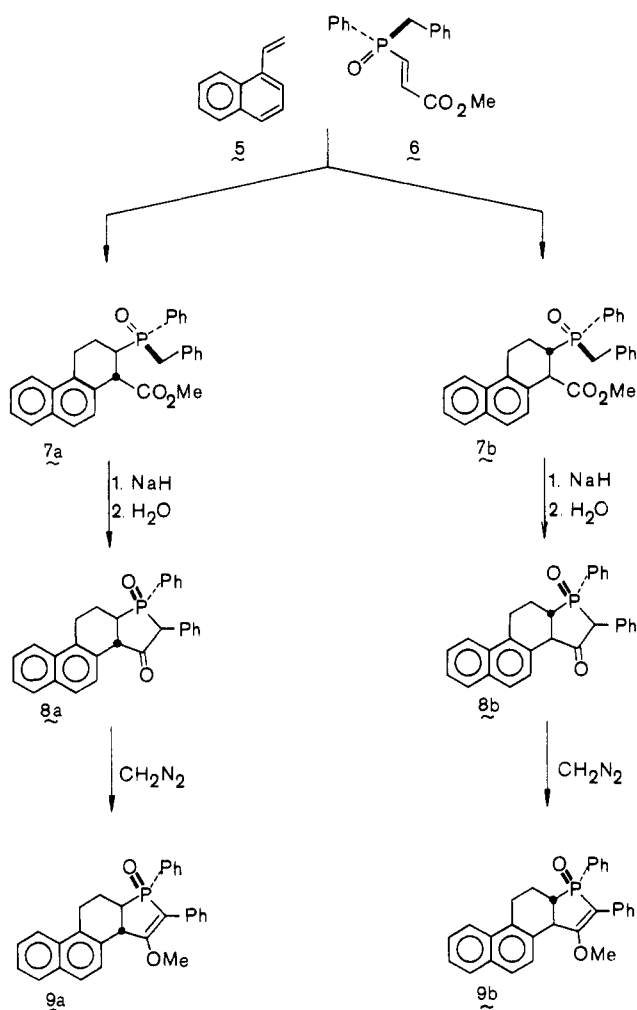
(14) Kashman, Y.; Awerbouch, O. *Tetrahedron Lett.* 1973, 3217.

(15) This type of reasoning was successfully applied in the case of dimethyl fumarate cycloadditions; see Walborsky, H. M.; Barash, L.; Davis, T. C. *Tetrahedron* 1963, 19, 2333.

Scheme I



Scheme II



such a model the predominant formation of **7a** calls for a preference of diene approach from the side of the benzyl phosphorus substituent. This is represented in Figure 1.¹⁷

Treatment of the adducts **7a** and **7b** with sodium hydride in tetrahydrofuran resulted in the formation of the phosphasteroids **8a** and **8b**, respectively, in ca. 80% yield in both cases. Both **8a** and **8b** are crystalline, high-melting solids (250 °C dec) and are poorly soluble either in common organic solvents or in water.¹⁸ Similar properties of many other 3-oxophospholane oxides have been observed^{13,19} and can be ascribed to the high enolic character of the 3-keto function and strong intermolecular hydrogen bonding.^{19,24}

(16) Overman, L. E.; Taylor, G. F.; Houk, K. N.; Domelsmith, L. N. *J. Am. Chem. Soc.* 1978, 100, 3182.

(17) Referee I is gratefully acknowledged for the helpful suggestion regarding Figure 1.

(18) Compounds **8a** and **8b** show increased solubility in alkaline water.

(19) (a) Quin, L. D.; Caputo, J. A. *Chem. Commun.* 1968, 1463. (b) Smith, D. G.; Smith, D. J. H. *Tetrahedron Lett.* 1973, 1249. (c) Quin, L. D.; Stokes, R. C. *J. Org. Chem.* 1974, 39, 686. (d) Purdum, W. R.; Berlin, K. D. *Ibid.* 1974, 39, 2904. (e) Former, K.; Henning, H. G. *Z. Chem.* 1974, 14, 477. (f) Martin, M.; Regitz, M.; Maas, G. *Justus Liebig's Ann. Chem.* 1978, 789.

To facilitate handling and preclude spectral ambiguities connected with keto-enol tautomerism, we transformed both phosphasteroids **8a** and **8b** into the respective enol ethers **9a** and **9b** by means of methylation with diazomethane in ether, in almost quantitative yields. **9a**: mp 266 °C; $[\alpha]_D^{20} +68.7^\circ$ (c 5.1, 10% CF₃COOH-CHCl₃); ³¹P NMR⁷ (10% CF₃COOH-CHCl₃) 51.2 ppm; ¹H NMR (10% CF₃COOH-CDCl₃) 3.51 (s, OCH₃), 4.24 ppm (dd, ³J_{H-H} = 7.7 Hz, ³J_{P-H} = 7.5 Hz, CH). **9b**: mp 259 °C; $[\alpha]_D^{20} +14^\circ$ (c 3.8, 10% CF₃COOH-CHCl₃); ³¹P NMR⁷ (10% CF₃COOH-CHCl₃) 51.7 ppm; ¹H NMR (10% CF₃COOH-CDCl₃) 3.49 (s, OCH₃), 4.45 ppm (t, ³J_{H-H} = 7.5 Hz, ³J_{P-H} = 7.5 Hz, CH).

The values of ³J_{H-C₁₃-C₁₄-H and ³J_{P-C₁₃-C₁₄-H are indicative of *trans*²⁰ and *gauche*²¹ nuclei, respectively, and suggest the assignment of *trans* C/D ring junctions in **9a** and **9b** and, therefore, in **8a** and **8b** as pictured in Scheme II.}}

In our opinion, the presented AB → ABC → ABCD_P approach to the enantiomeric phosphasteroids has features of a general synthetic method since a number of bicyclic dienes and a series of dienophiles with different substitution patterns are likely to compose a suitable pair of reactants. The application of 1-vinyl-6-methoxy-3,4-di-



hydronaphthalene (**10**) and the phosphorus dienophile **11** for the syntheses of closer analogues of naturally occurring steroids is currently in progress.

Registry No. 5, 826-74-4; 6, 81027-60-3; **7a**, 81075-98-1; **7b**, 81075-99-2; **8a**, 81027-61-4; **8b**, 81076-00-8; **9a**, 81129-82-0; **9b**, 81027-62-5.

(20) A similar HCCH coupling of 7 Hz was observed for the closely related *trans*-1-benzyl-2-phenyl-4,5-dimethylphospholan-3-one 1-oxide system;^{19d} for dihydrothiophene ³J_{HCCH}(*trans*) = 7.5 Hz was found; see Abraham, R. J.; Sheppard, R. C.; Thomas, W. A.; Turner, S. *Chem. Commun.* 1965, 43.

(21) The value of 28 Hz for ³J_{PCC} coupling was obtained for antinuclei in a rigid system;²² the corresponding "averaged" value for triethylphosphine oxide is 16.3 Hz.²³

(22) Wetzel, R. B.; Kenyon, G. L. *J. Am. Chem. Soc.* 1974, 96, 5189.

(23) Hendrickson, J. B.; Maddox, M. L.; Sims, J. J.; Kaesz, H. D. *Tetrahedron* 1964, 20, 449.

(24) For a general discussion, see: Quin, L. D. "The Heterocyclic Chemistry of Phosphorus"; Wiley-Interscience: New York, 1981; Chapter 2, pp 62-64.

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